

Lectures 8 -11: Metabolism

I. The breakdown of glucose to pyruvate

A. Glycolysis ()

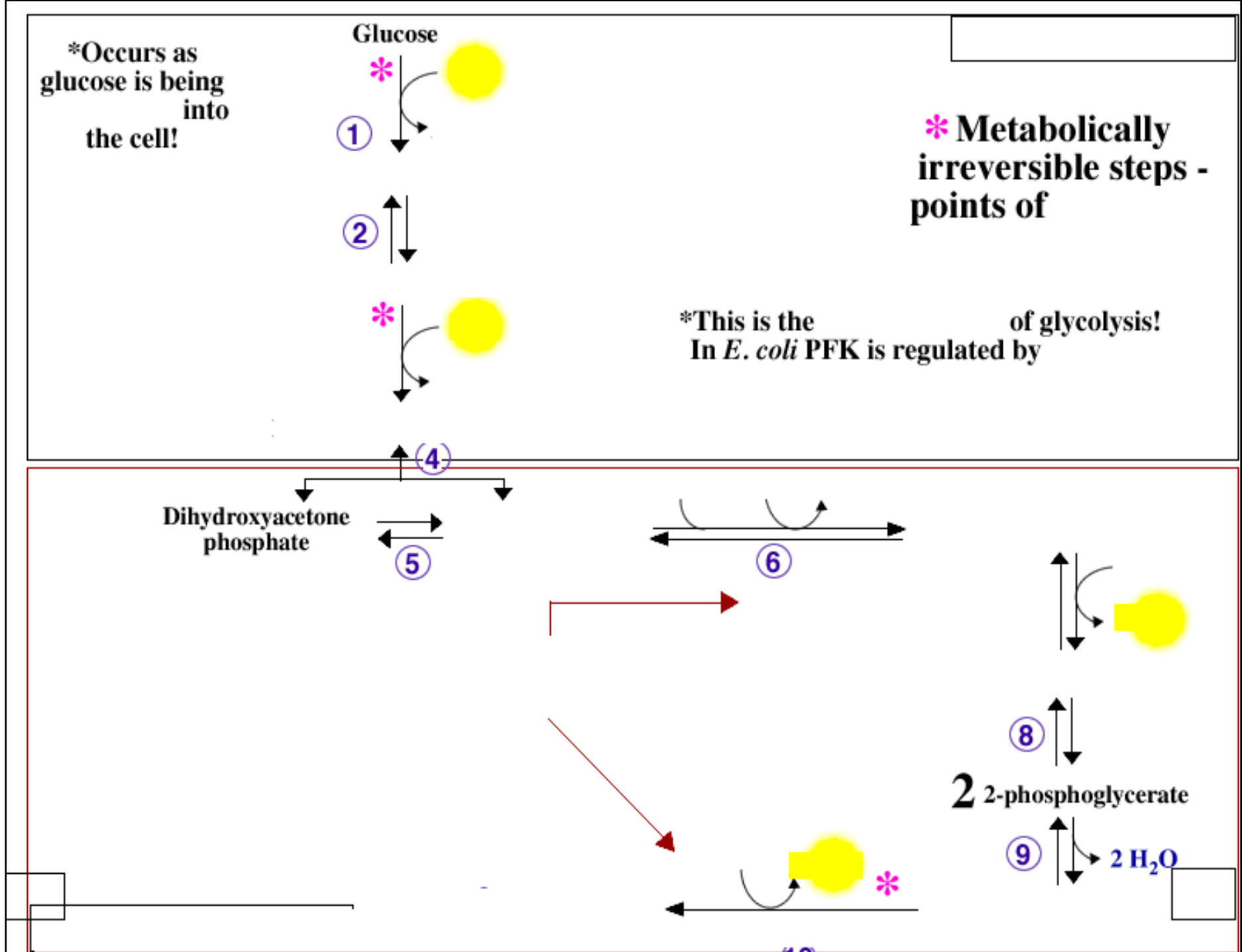
1. Facts

a. Glycolysis is a pathway that occurs in the of both prokaryotic and eukaryotic cells.

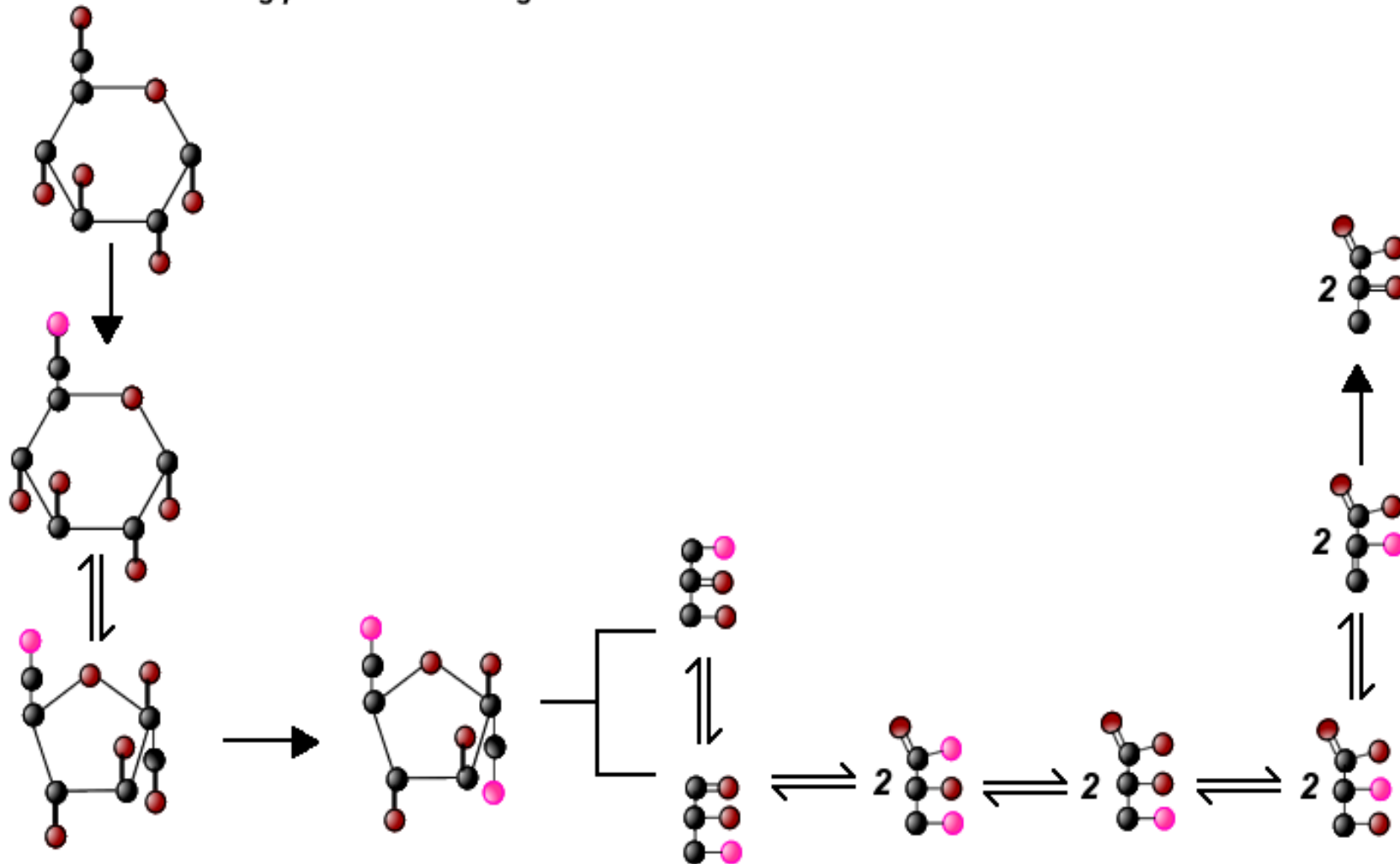
b. Glycolysis can occur under conditions.

c. Glycolysis is an pathway.

2. Reactions



Practice labeling the structures and the stages of glycolysis. Also practice labeling the ATP and reducing power consumed/generated.



If 3.34×10^{21} molecules of glucose are oxidized by a culture of bacteria living on a GSA plate, how many net ATP molecules are harvested via substrate level phosphorylation in the Embden-Meyerhof Pathway? How much reducing power is created (# of reduced NADH molecules)?



B. The Pentose Phosphate Pathway

1. Functions

a. to form ribose 5-phosphate which is needed for the

b. to form $NADPH$, whose electrons are needed in reductive biosynthesis (e.g. the synthesis of fats).

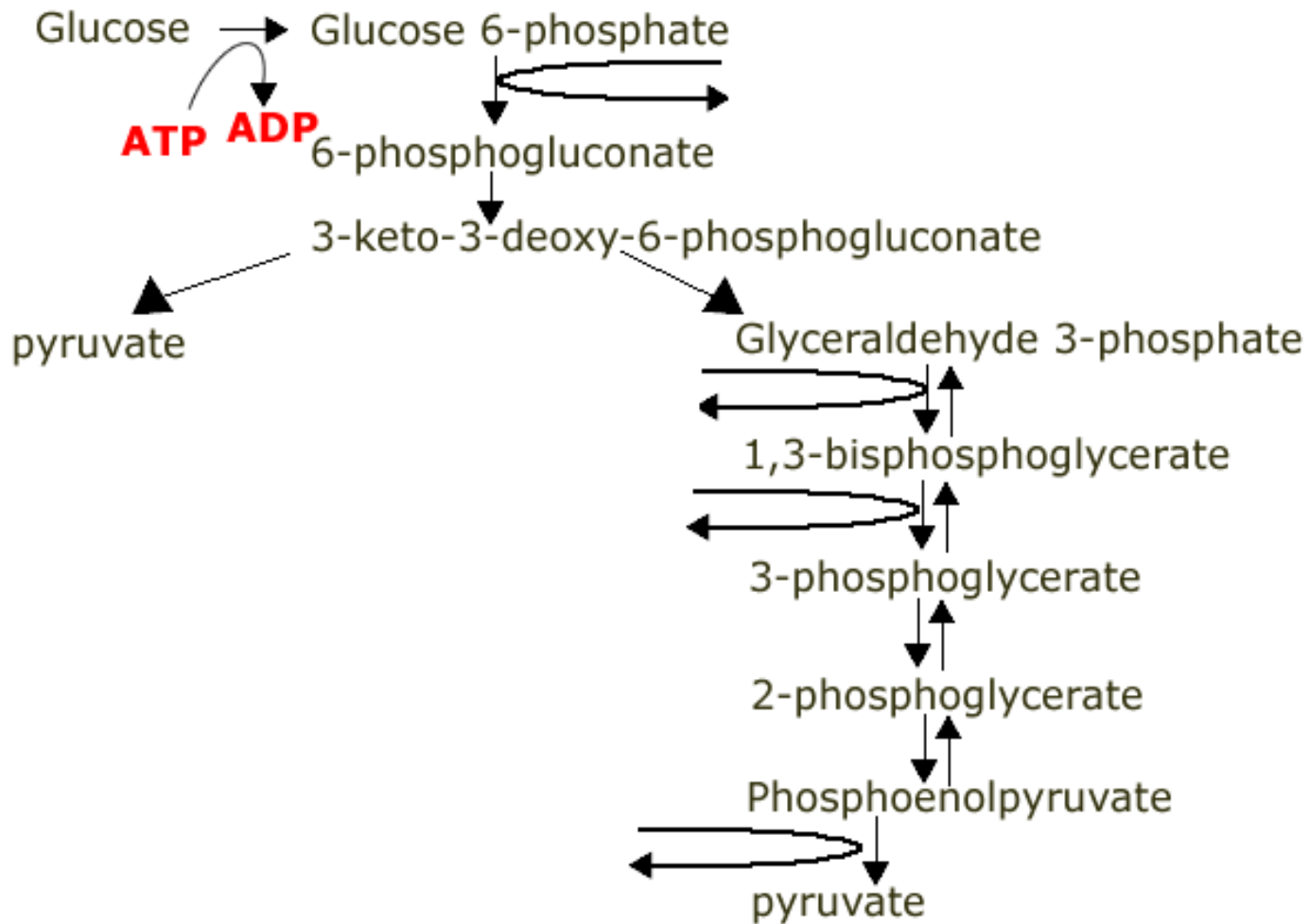
2. Occurs in the cytosol of prokaryotic and eukaryotic cells.

3. Can operate at the cytosol as the Embden-Meyerhof Pathway.

4. Can operate

C. The Pathway

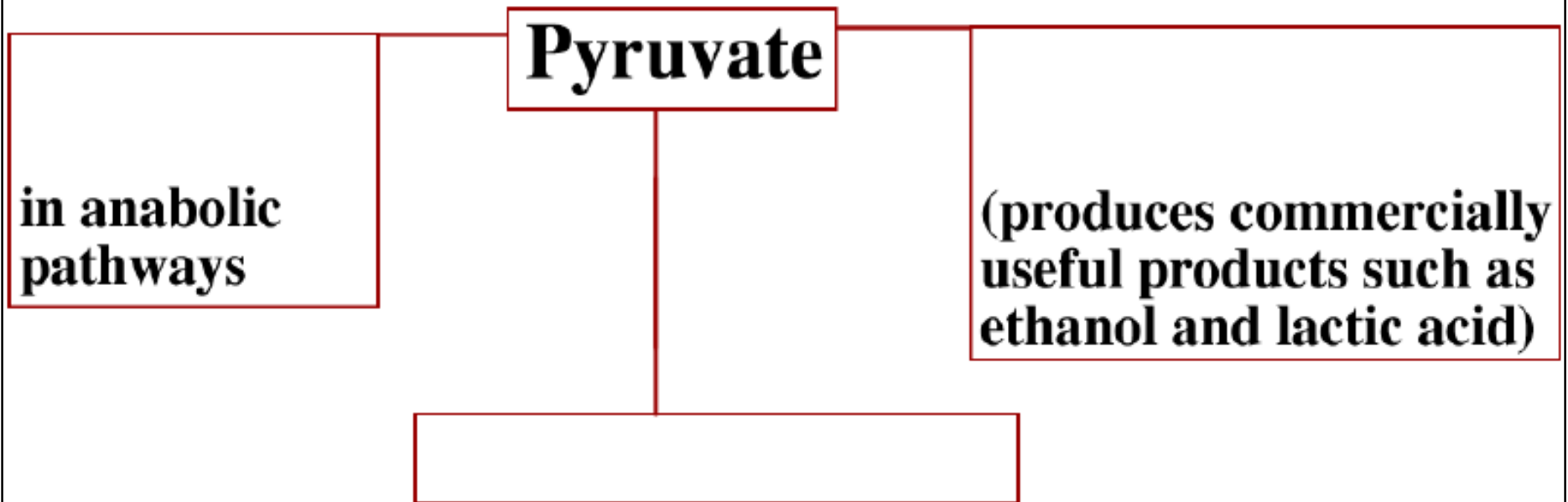
1. An alternative pathway that converts hexose sugars to pyruvate. It is primarily observed in (e.g. *Pseudomonas*, *Rhizobium*, *Azotobacter* and *Agrobacterium*).
2. The steps of the triose stage are the same as those in (Embden-Meyerhof pathway)
3. The steps of the hexose stage are similar to those in the pathway



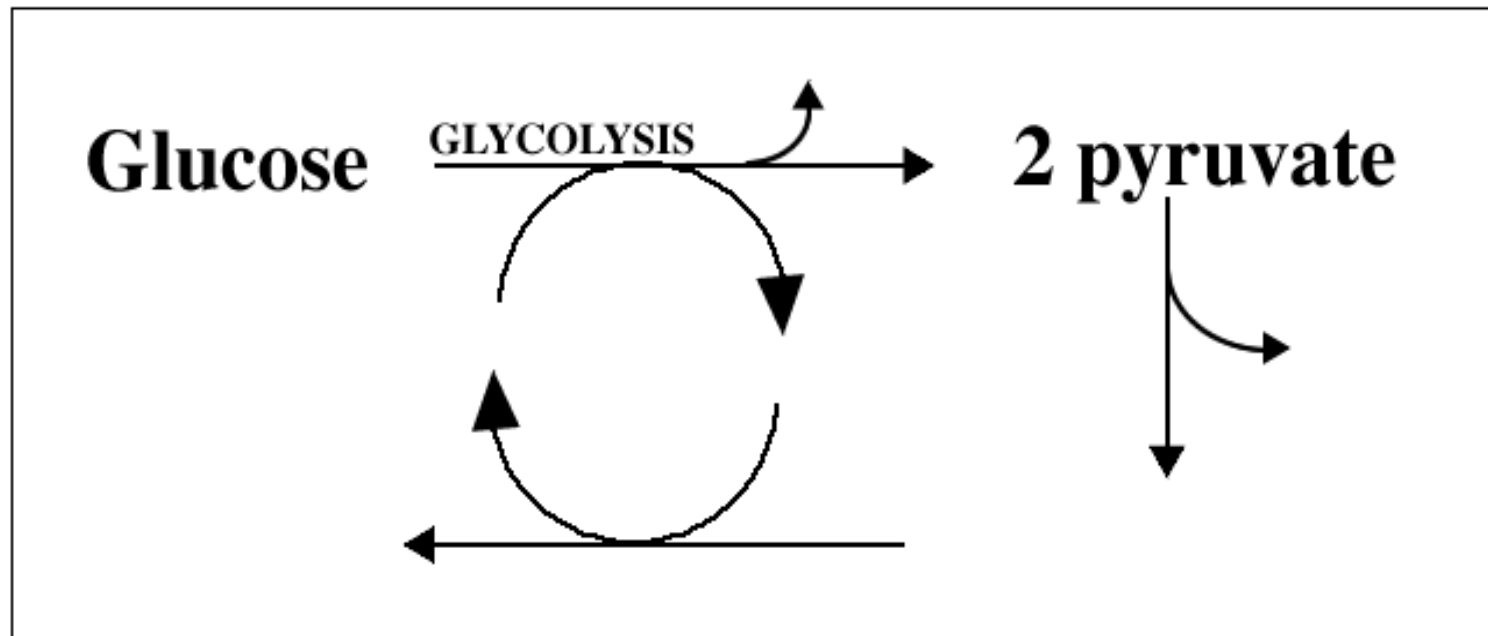
4. The Net Yield is
glucose metabolized.

and per

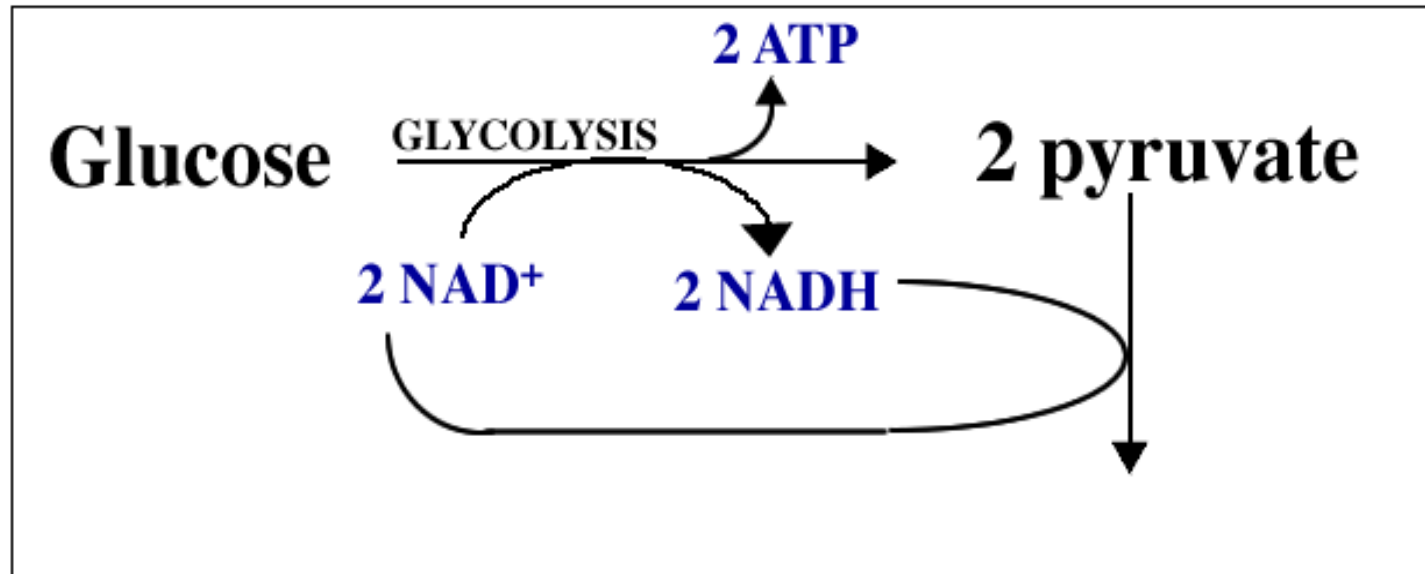
II. Fermentation - A potential fate of pyruvate



A. Alcoholic fermentation - Used by many fungi, some bacteria and protists



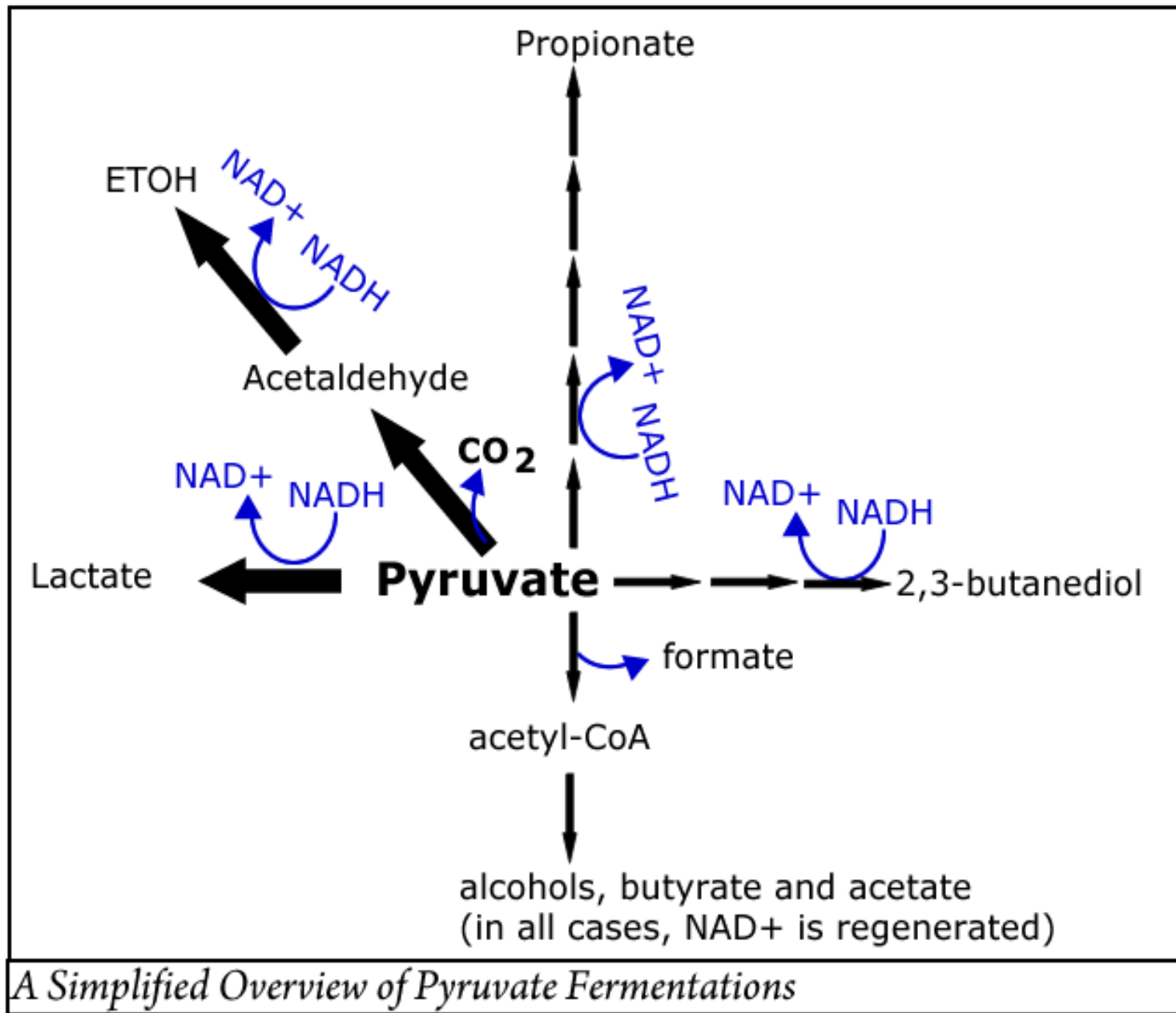
B. Lactic acid fermentation - Used by bacteria (the lactic acid bacteria / *Bacillus*), and some protists



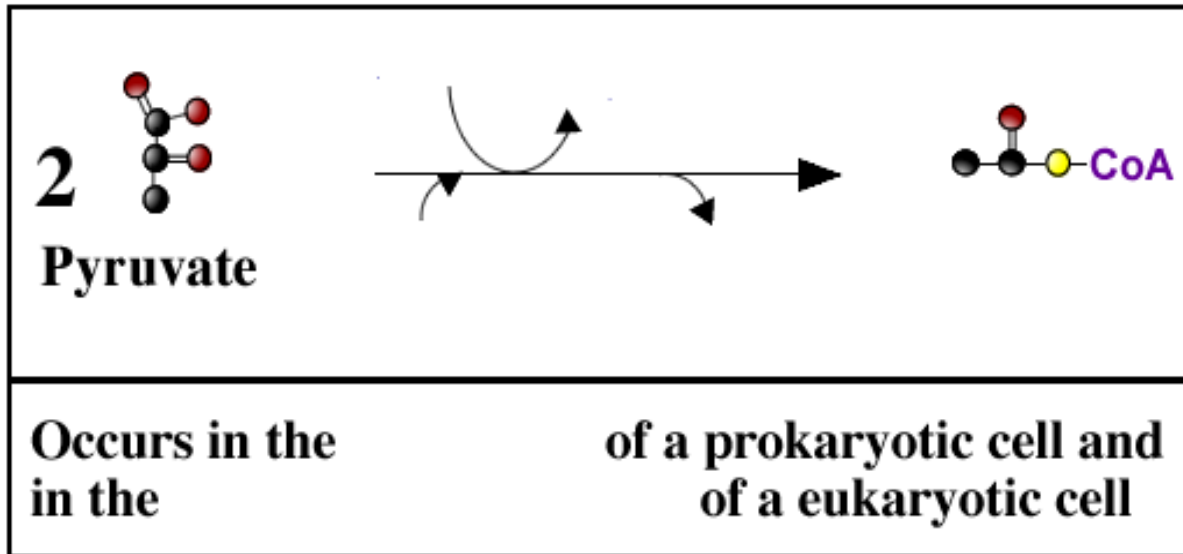
*Some bacteria produce only lactic acid as a product of fermentation () whereas others produce ethanol and CO₂ in addition to lactic acid ().

*The purpose of fermentation is to
 reduced during glycolysis. This allows
 glycolysis to continue to run and ATP to continue to
 be generated

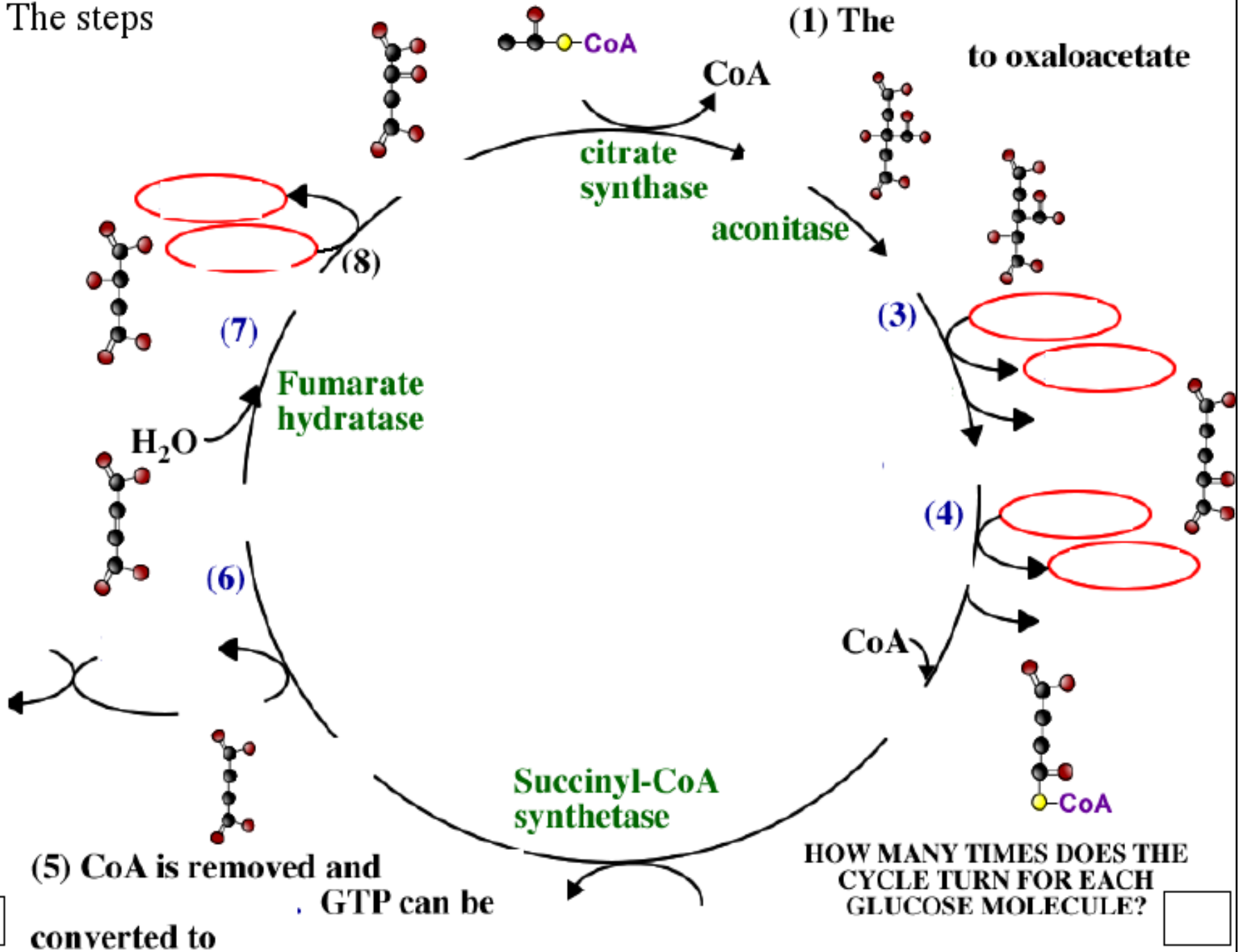
(e.g. O₂)!



III. The Transition Step (FATE = FURTHER OXIDATION)



B. The steps



C. The net yield of the TCA cycle

1. ATP

ATP/acetyl or ATP/glucose in step 5

2. Reducing power

NADH/acetyl or NADH/glucose in steps 3,4 and 8

FADH₂ (QH₂)/acetyl or FADH₂ (QH₂)/glucose and in step 6

3. Precursor metabolites

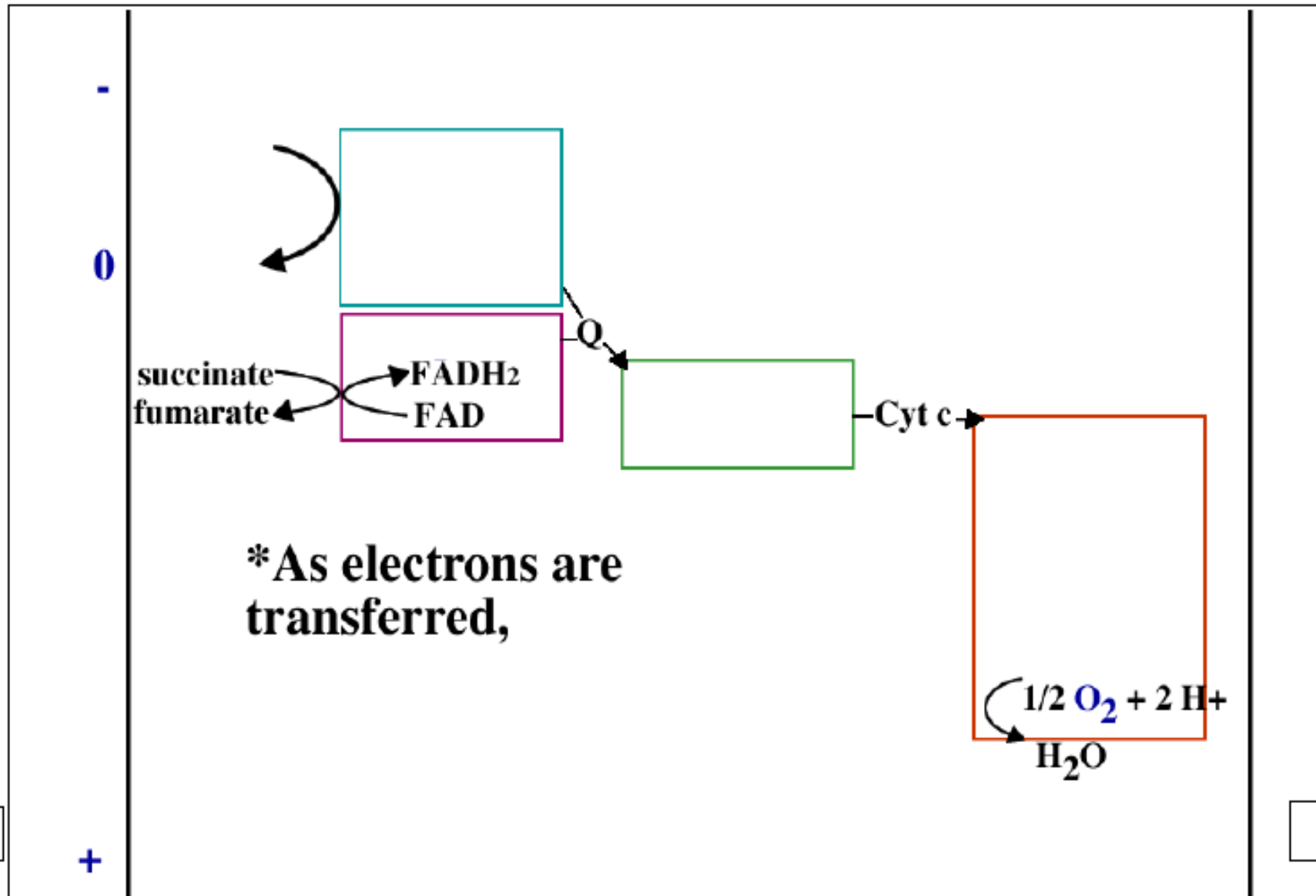
(α -ketoglutarate, succinyl-CoA and Oxaloacetate are common building blocks in biosynthetic reactions)

If 1.2×10^{18} molecules of glucose are catabolized, how many ATP molecules are net via substrate-level phosphorylation during glycolysis and the TCA cycle?



V. The Electron Transport Chain and Oxidative Phosphorylation

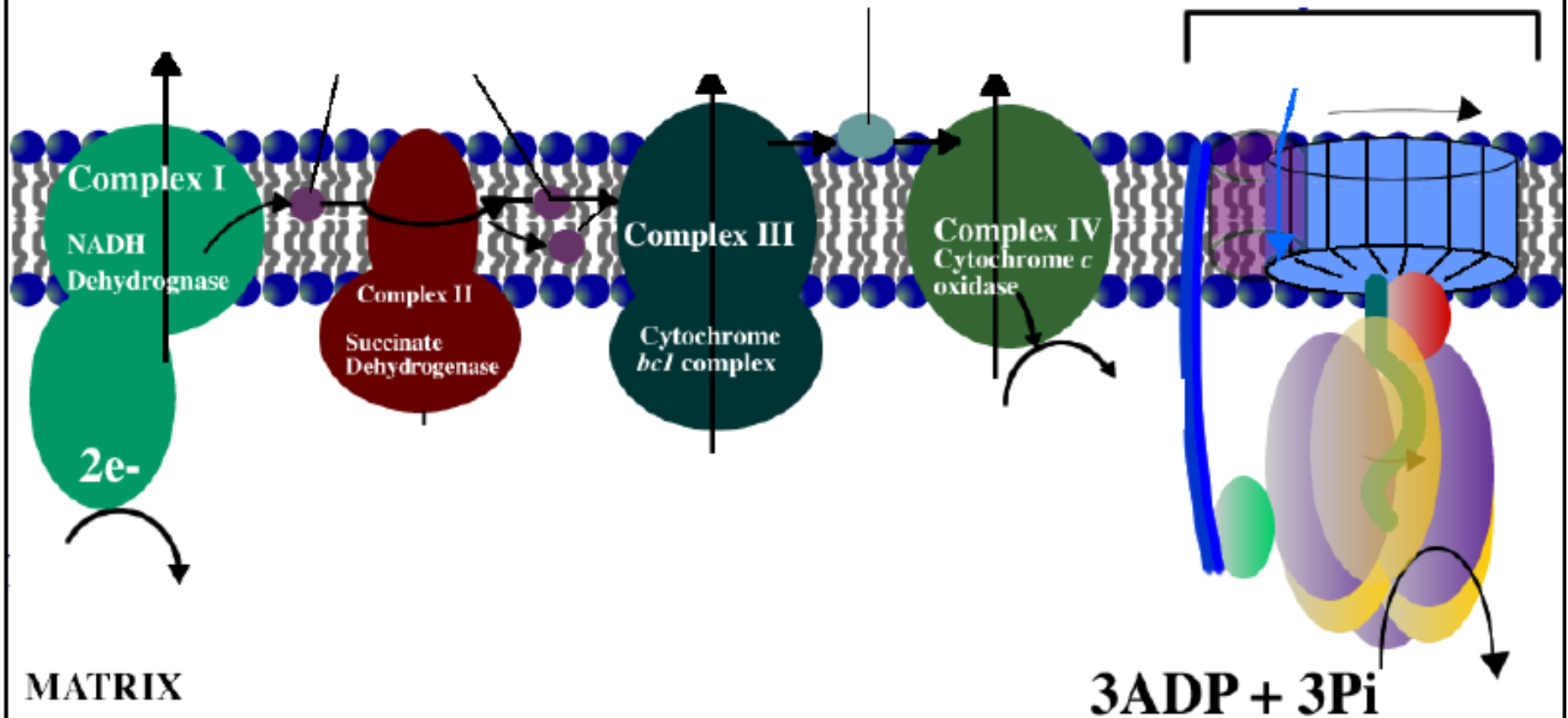
A. Electrons are transferred from NADH or from the oxidation of succinate to one of the membrane-embedded carriers. They are then transferred along the entire electron transport chain.



B. ATP Synthase and the Chemiosmotic Hypothesis

The Electron Transport Chain of Mitochondria

INTERMEMBRANE SPACE



How many protons are pumped upon the complete oxidation of NADH ? FADH_2 ?

A proton concentration gradient (PMF) serves as an E reservoir for driving

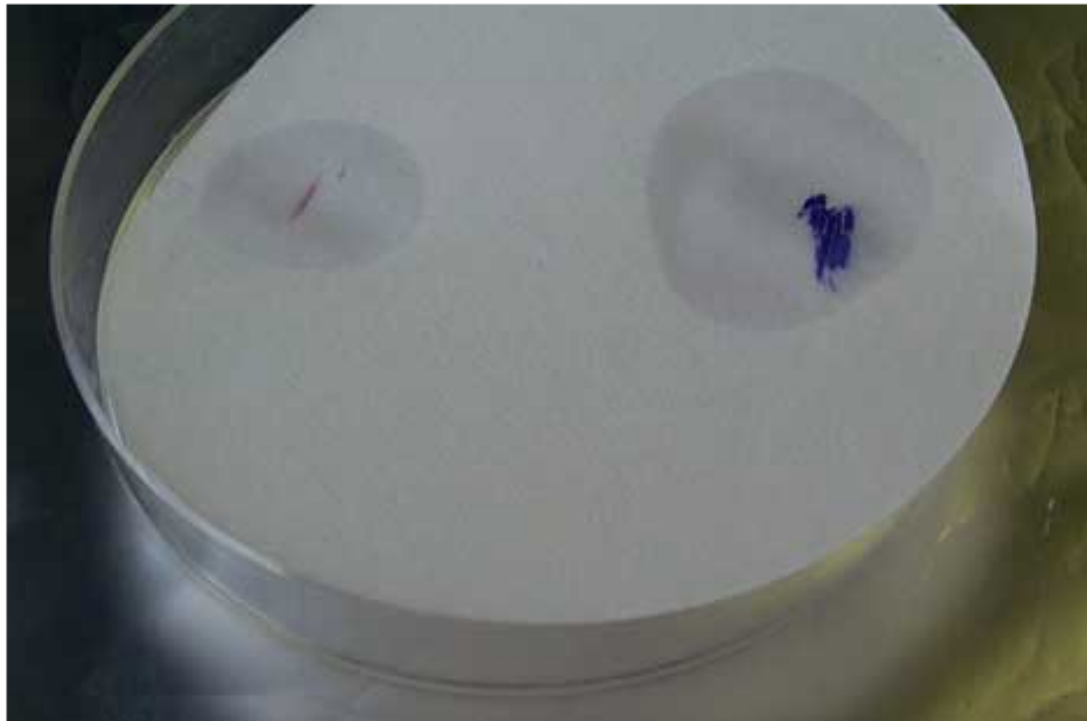
C. The components of the electron transport chains of prokaryotes

- *Pseudomonas* and *Neisseria* have

(Activity of this complex can be used to help members of these genera.)

- *E. coli* has two different

alternative complexes allowing it to use a variety of energy sources and (Fig. 10.12 (8th) and 11.13 (9th))



Theoretical ATP yield from aerobic respiration of glucose in eukaryotes:

Conversion factors

For each pair of electrons delivered to the ETC by NADH,
are formed via oxidative phosphorylation.

For each pair of electrons delivered to the ETC by $\text{FADH}_2/\text{QH}_2$,
are formed via oxidative phosphorylation.

Metabolic process	# of ATP via substrate-level phosphorylation	# of ATP via oxidative phosphorylation	Total
Glycolysis			
The Transition Step			
The TCA Cycle			
Total			

*Depending on how the electrons from NADH enter the mitochondrion,
thus affecting the overall total.

Just as the composition of bacterial ETCs varies, so to does the

VI. Anaerobic respiration

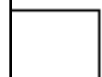
Organisms capable of anaerobic respiration can use _____ as the terminal electron acceptor.

A. Some facultative anaerobes, like *E. coli* use _____ as the final electron acceptor:

B. Some anaerobes called sulfate-reducers use _____ as the terminal electron acceptor.

C. Other anaerobes, such as the methanogens, use _____ as their terminal electron acceptor.

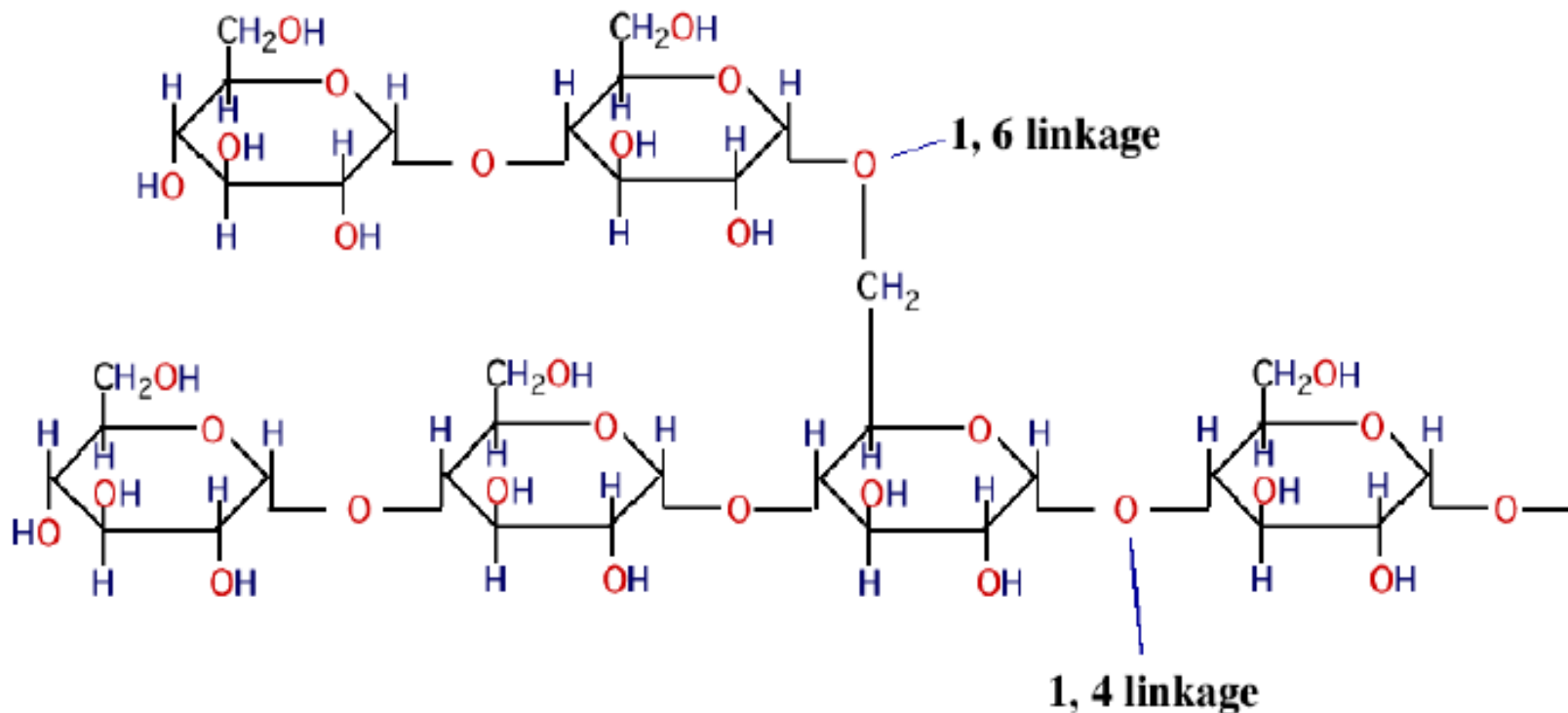
D. Anaerobic respiration produces _____ than aerobic respiration.



VII. Catabolism of carbohydrates, triacylglycerols and proteins

A. Carbohydrates

1. Monosaccharides besides glucose () can enter the glycolytic pathway (Fig 10.20 (8th) or 11.21 (9th)).
2. Common can be cleaved to monosaccharides which can then enter the glycolytic pathway.
3. Starches are digested by



4. Cellulose is digested by enzymes called . Only a few

microorganisms such as and some

produce these enzymes.



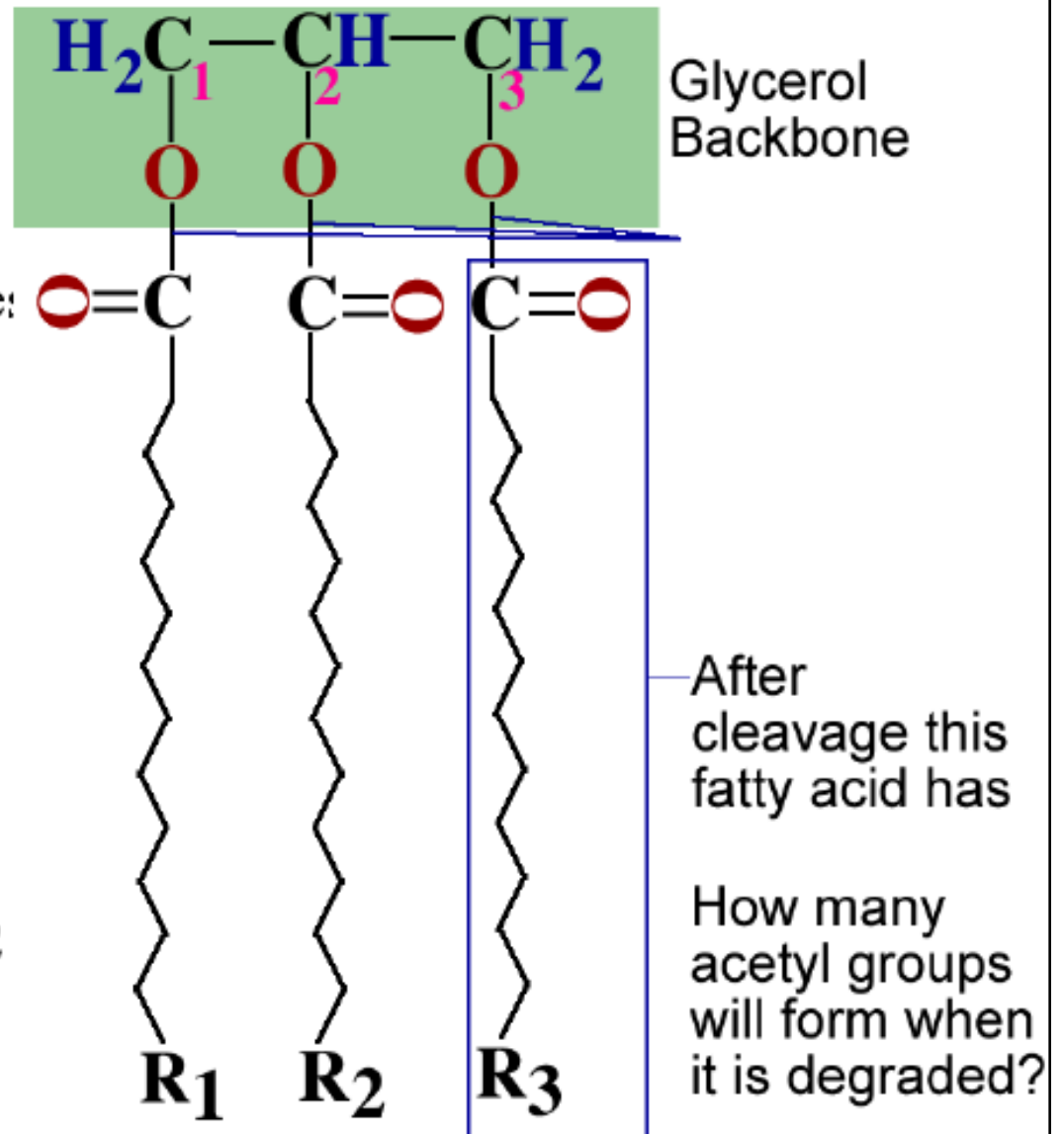
-The glucose subunits released when polysaccharides are digested enter the

5. Some bacteria even produce enzymes that degrade pesticides and other pollutants (), transforming these

compounds into normal metabolic intermediates

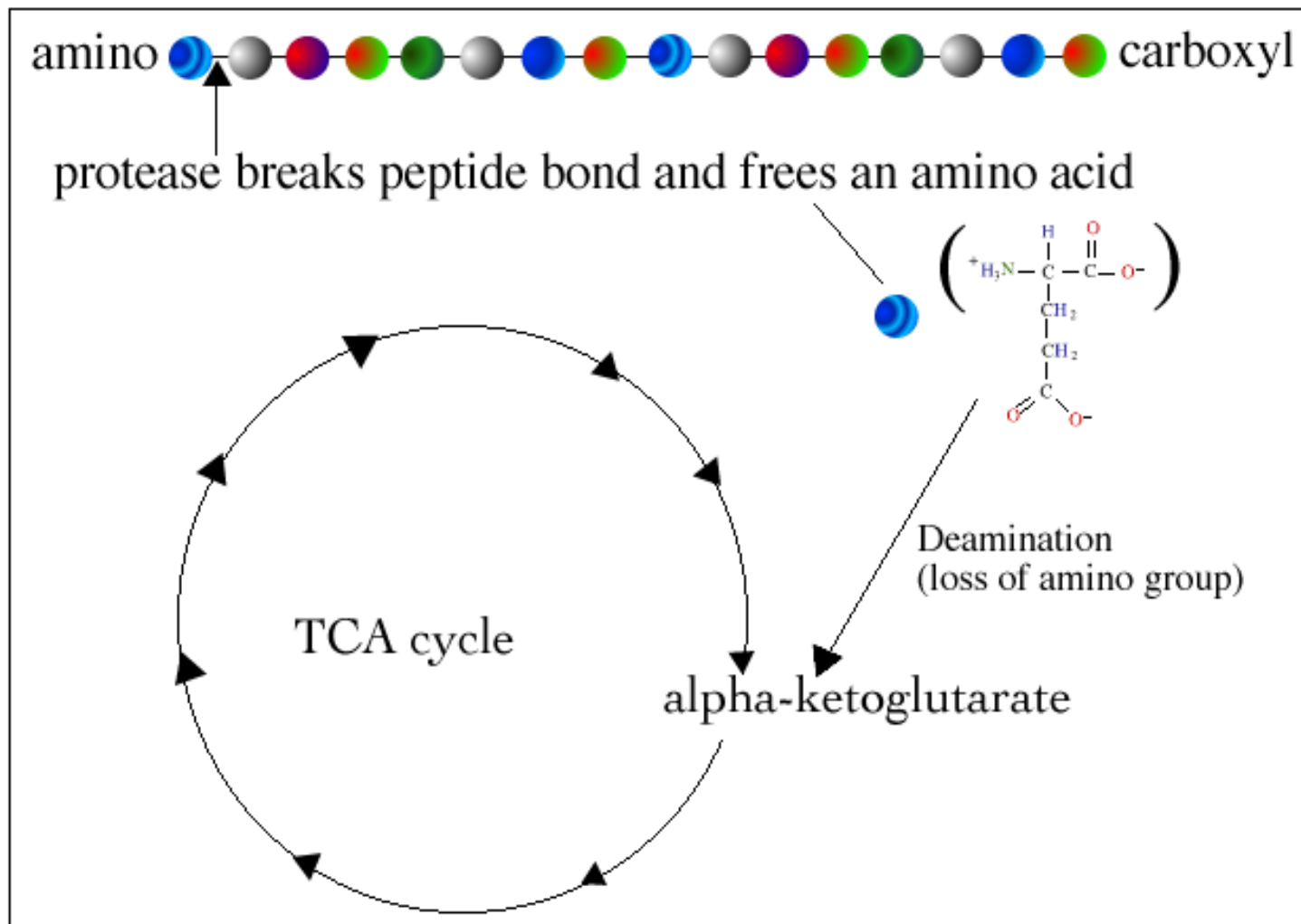
B. Lipids

1. Fats (triglycerides) are hydrolyzed by secretory enzyme: called .
2. The glycerol backbone is converted into
3. The fatty acids are degraded 2 carbons at a time to form



C. Proteins

1. Enzymes called break the peptide linkages.
2. A deamination reaction leaves just the carbon skeleton which can enter the



VII. Chemoautotrophs (chemolithotrophs)

A. Prokaryotes are alone in their ability to use
as an energy source.

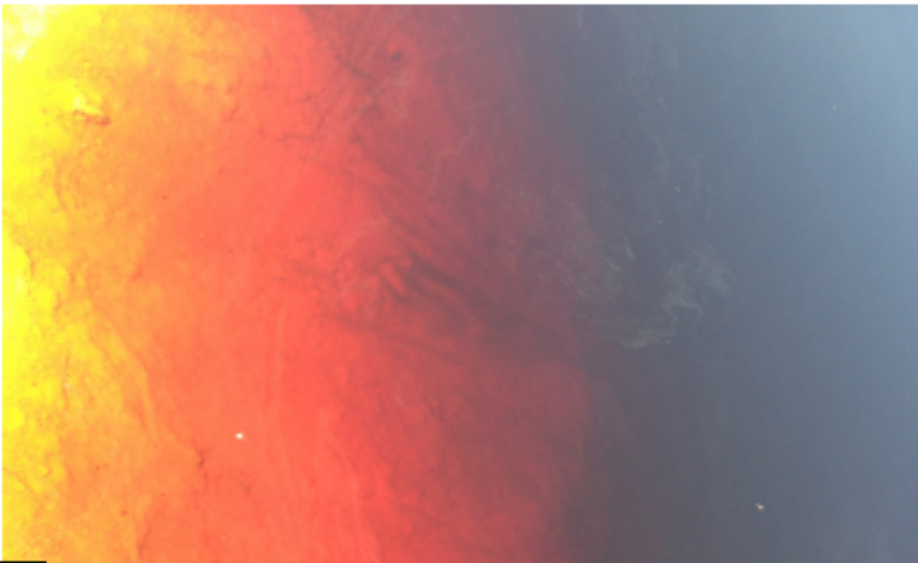
Many of these inorganic compounds are byproducts of

Type of chemoautotroph	Identity of chemical oxidized for energy
bacteria	hydrogen () gas
bacteria	hydrogen sulfide ()
bacteria	reduced forms of iron ()
bacteria type 1 type 2	(NO ₂ ⁻) ammonia (NH ₃)

*electrons extracted during the oxidation of these chemicals are carried to an
electron acceptor. Generally serves as the terminal

B. Chemoautotrophs often thrive in where
are often found.

Table 10.4 (8th) or 11.5 (9th)



Acid mine drainage
Photos taken by
Rachel near her
hometown of
Leadville

