



Low-Pressure Methanol Production from CO₂ Using Catalysts

UW ID: 17-002

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Patent Status:

Patent Pending

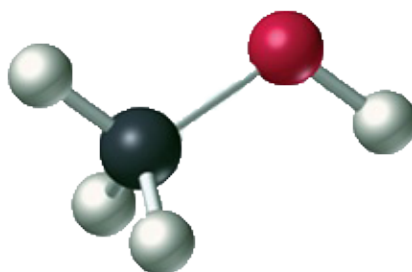
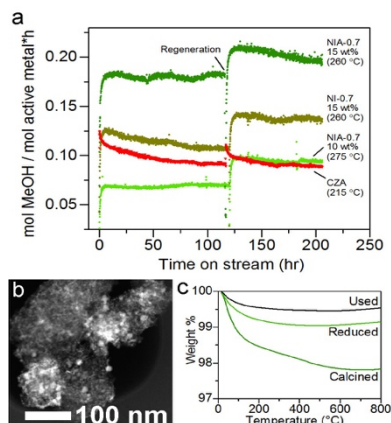
Description of Technology

Increased atmospheric carbon dioxide levels have motivated research on CO₂ capture and how it can be used for beneficial purposes. Methanol (MeOH) is an organic compound used in alternative fuels and is in high demand globally. In the past, methanol has been synthesized from CO₂ directly and indirectly through a CO conversion process by using a reverse water-gas shift reaction, but this is usually optimized for high-pressure conditions using CO₂ and CO. There has been a need for an alternative catalyst to provide a more effective way to use CO₂.

Researchers at the University of Wyoming have developed a methanol synthesis catalyst that is designed to effectively utilize CO₂ for methanol synthesis at a low pressure with high stability and selectivity, without requiring precious metals. The new composite catalysts contain nickel and indium as co-catalysts and are synthesized via a phyllosilicate precursor. The metals used in this process have advantages in both their catalytic and structural attributes.

Applications

- The required H₂ can be obtained using renewable energy resources such as biomass and solar based water splitting. The methanol produced by this process is renewable and has the potential to be a fossil fuel replacement.
- Low pressure CO₂ and H₂ based MeOH synthesis catalysts means synthesis that is not only less energy-intensive, but can also be beneficial to development of other energy technologies such as direct methanol fuel cells



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